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Original Research Article



Potential of Nanoencapsulated *Curcuma longa-Andrographis paniculata* as an Anticancer Agent

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ABSTRACT

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Anticancer agents can inhibit and destroy cancer cells by various mechanisms. Some plants are reported to have bioactive components that can be developed as anticancer agents, such as *Curcuma longa* and *Andrographis paniculata*. Although herbal-based drugs have good potential, they have limitations when applied in vivo. Developing drug delivery systems is a potential solution by utilizing nano *C. longa-A. paniculata* as anticancer agents. The nanoencapsulation of *C. longa-A. paniculata* with carboxymethyl chitosan was performed using the ultrasonication method. The characterization of nano *C. longa-A. paniculata* (FB-NPs) included physicochemical properties, FTIR, AFM, stability, loading, and release of its bioactive components. FB-NPs had a particle size of 222.60 ± 0.87 nm. The loading amount and efficiency of FB-NPs were 17.83 ± 3.04% and 35.66 ± 6.08%, respectively. FB-NPs showed greater stability under changes in pH, temperature, and salt concentration compared to the free *C. longa-A. paniculata* formula (FB). The anticancer activity of FB-NPs against the MCF7 cell line showed stronger inhibition, with an EC₅₀ value of 13.71 ± 2.66 µg/mL compared to FB. The nanoencapsulation process of *C. longa-A. paniculata* with carboxymethyl chitosan has potential as an herbal-based anticancer agent, which can significantly contribute to the development of breast cancer treatment.

Keywords: Andrographis paniculata, Anticancer, Carboxymethyl chitosan, Curcuma longa, Nanoencapsulation.

Introduction

Breast cancer is one of the most frequently diagnosed malignant in women, with an estimated 2.3 million cases worldwide.^{1,2} Breast cancer is one of the most studied cancers, with a curable rate of 70.00-80.00% in patients with early-stage and non-metastatic disease. However, it still has high mortality rates in several countries, and a fiveyear estimation for mortality risk is still used in the recent diagnosis.³ This cancer is a heterogeneous disease that has molecular features, including activation of human epidermal growth factor receptor 2 (HER2), activation of hormone receptors (estrogen receptor and progesterone receptor), and breast cancer gene (BRCA) mutation. Several risk factors for breast cancer include sex, age, reproductive and hormonal status, molecular alterations, genetic predisposition, ionizing radiation, diet, lifestyle, and obesity.^{1,4} Treatment options for breast cancer vary depending on the type and the stages of cancer, which include surgery, radiotherapy, immunotherapy, neo-adjuvant therapy, endocrine therapy, etc.^{5,6} Most of these treatments intervene in various stages of cancer development, from the single cell to initiation stage promotion, migration, and progression.7 Several medicinal herbs have been heavily researched for their anticancer potential due to their uniqueness compared to synthetic compounds or molecules, such as curcumin and green chiretta.8,9

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Curcumin is a bioactive component of turmeric obtained from the Curcuma longa plant, a member of the Zingiberaceae ginger family widely found in Indonesia. It is commonly used in food flavoring, coloring, and traditional medicine.^{10,11} It is known to have antioxidant, antibacterial, anti-inflammatory, and anticancer properties. It is observed to have highly cytotoxic anticancer activity against breast and prostate cancer, inhibiting cell proliferation and triggering apoptosis.12-¹⁴ Meanwhile, Andrographis paniculata is known to play a role in lowering fever, reducing inflammation, inhibiting the formation of atherosclerosis, and inhibiting the formation of reactive oxidative species (ROS).^{15,16} In addition, previous studies have shown that the constituent of A. paniculata was found to have antitumor activity.17 Encapsulating these anticancer agents into biocompatible and biodegradable nanoparticles (NPs) could improve the efficiency and effectiveness of their chemoprevention activity ^{18,19}. Furthermore, it may overcome their poor adsorption, low systemic bioavailability, gastrointestinal hydrolysis, and short half-life.²⁰ Chitosan as the shell for anticancer agents could be an option. The nanoencapsulation process can enhance anticancer activity, as seen in nanoencapsulated Syzygium polycephalum with carrageenan-folate, which showed increased inhibitory activity against T47D and HeLa cell lines.²¹ Carboxymethyl chitosan (CMCS) is a derivative of chitosan that is more hydrophilic than chitosan. It consists of active hydroxyl (-OH), carboxyl (-COOH), and amine (-NH2) groups that enhance its solubility in neutral pH.²² CMC has potential as a carrier for various biological active agents due to its ability to control the drug release rate and high biocompatibility.^{22,23} Previously, CMC has been observed to have anticancer and antitumor properties as a single molecule and conjugate of anticancer drugs.^{22,24,25}This study synthesized and characterized the formulation of C. Longa-A. Paniculate encapsulated with chitosan to investigate its anticancer potential against breast cancer. Physical and chemical characterization, as well as in vitro tests, were performed.

Materials and Methods

Material

C. longa rhizome, *A. paniculata* leave, HCl (Merck), NaOH (Merck), methanol, aquades, carboxymethyl chitosan, KCl (Merck), KH₂PO₄ (Merck), K₂HPO₄ (Merck), H₃BO₃ (Merck), curcumin (Sigma-Aldrich), quercetin (Sigma-Aldrich), DPPH (2,2-Diphenyl-1-picrylhydrazyl) (Sigma-Aldrich), NaCl (Merck), ascorbic acid (Sigma-Aldrich).

Sample Preparation

Rhizome of *C. longa* L. (certificate No. 002/03.542.Jl.Genbinesia) and leaves of *A. paniculata* (Burm.f.) Wall. Ex Nees (certificate No. 002/03.543.Jl.Genbinesia) taken from Mount Lawu, Jogorogo, Ngawi, East Java, Indonesia (7°33'46''S 111°13'23''E). Samples of *C. longa* rhizomes and *A. paniculata* leaves were dried and grounded, respectively. A total of 1 kg each was extracted with methanol for 24 hours and repeated thrice. The extract of each sample was concentrated using a rotary vacuum evaporator (BUCHI R-100).

Herb formulation

The herb formula consists of *C. longa* and *A. paniculata* extract with ratio 3:1 and antioxidant bioactivity screening using the DPPH method.²⁶ The best results are selected to continue with the following process.

Nanoencapsulation of C. longa-A. paniculate

The nanoencapsulation process of *C. longa-A. paniculata* herb formula refers to the previous research.²⁷ Nano herbs *C. longa-A. paniculata* were obtained from the nanoencapsulation process with carboxymethyl chitosan. Zeta potential, polydispersity index (PDI), and particle size were the three physicochemical characteristics of nano *C. longa-A. paniculata* that were determined by Dynamic Light Scattering (DLS, Zetasizer Nano ZS, Malvern). Using FTIR (Shimadzu IRTracer-100), the functional groups of the nano *C. longa-A. paniculata* were identified. Simultaneously, Scanning Electron Microscope (SEM, Zeiss EVO MA 10) analysis was conducted on the morphology of nano *C. longa-A. Paniculate*.

Loading and release of nano C. longa-A. paniculate

Nanoencapsulation is an application of nanotechnology that can improve stability, and bioavailability of bioactive compounds because nano-size can increase the surface area. Stability tests were performed based on the pH, salt concentration, and temperature. The test results were conducted by observing changes in the UV-Vis spectra (Shimadzu UV-1800) and turbidity levels using a turbidimeter.^{27,28} The release study was conducted at pH 2, 7, and 8.5 (Eq. 1). The loading amount (%LA) and efficiency (%LE) calculations are determined using Eq. 2 and 3.

$$Ct^{1} = Ct^{0} + \frac{\nu}{v} \sum_{0}^{i-t} Ct \dots(1)$$

Where Ct¹: concentration correction at t time

Ct⁰: measured concentration at t time

- V : Total volume of buffer used
- v : volume of aliquots

$$\% LA = \frac{\text{Mass of samples on FB-NPs}}{\text{Mass of FB-NPs}} x100\% \dots (2)$$

$$\% LE = \frac{\text{Mass of samples on FB-NPs}}{\text{Mass of samples in feed}} x100\% \dots (3)$$

Toxicity assay

MCF7 cell lines were cultured in a 96-well plate and then incubated at 37°C with 5.00% CO₂ until the cell growth percentage reached 70.00%. Cells were treated with each variation of formulation and incubated at 37°C with 5.00% CO₂ gas for 48 hours. A number of 100 μ L of each sample and positive control of the microtube into each 96-well plate that already contains cells. About 9 mL of media in a tube to which 1 mL of "Resazurin Sodium Salt-Powder, BioReagent" (10 μ L reagent for 90 μ L medium), then 100 μ L the mixture solution into each well of the microplate and then incubate for 1-2 hours until a color change is visible. Resazurin Sodium Salt reagent in a living cell will be reduced from the resazurin blue compound without intrinsic fluorescent value,

being a red and very fluorescent resorufin compound. The conversion value is proportional to the number of active metabolic cells and can be measured quantitatively. The absorbance was measured at a wavelength of 570 nm using a multimode reader (TECAN Infinite M200 Pro).

Statistical analysis

Statistical analyses were performed using GraphPad Prism 10. p-value of the data were analysed using paired Student's t-test with two-tailed distributions. A p-value < 0.05 was considered statistically significant.

Results and Discussion

Phytochemical screening for secondary metabolites

C. longa-A. paniculata Herbal Formula

Herb formula of *C. longa-A. paniculata* was synthesized with various compositions, and its antioxidant activity formula was measured. Overall results show FA has the best antioxidant activity compared to FB and FC. The IC₅₀ values of each formula were $308.02 \pm 16.17 \mu$ g/mL, $400.53 \pm 49.21 \mu$ g/mL, and $1699.14 \pm 152.19 \mu$ g/mL (Figure 1). However, during the nanoencapsulation process, FA and FC were observed to be unstable compared to FB. Therefore, FB was used for nanoencapsulation and anticancer activity testing, while *C. longa-A. paniculata* was formulated and its nano properties evaluated (FB and FB-NPs) containing 2.16 mg curcumin/g *C. longa* extract and 0.06 mg quercetin/g *A. paniculata* extract. The bioactive component level is used as the base of loading-release data formulation.

Characterization of nano C. longa-A. paniculata

C. longa and A. paniculata were formulated as nano herbs encapsulated with carboxymethyl chitosan. The characterization of FB-NPs consisted of particle size and polydispersity index (PDI) (Table 1). Particle size ensures that each sample is classified as a nanoparticle. Nanoparticles are colloid-sized particles with a maximum diameter of 500 nm.²⁹ The particle sizes of FB and FB-NPs are 571.90 \pm 65.09 nm and 222.6 \pm 0.87 nm, respectively. The polydispersity index (PDI) measures sample heterogeneity based on particle size caused by size distribution, aggregation, and agglomeration in the sample. The PDI of FB and FB-NPs are observed at 0.56 ± 0.10 and 0.24 ± 0.018 , respectively (Table 1). These results indicate that the size distribution of FB and FB-NPs tends to be homogeneous. PDI values less than 0.05 (mono dispersion sample) indicate a narrow particle size distribution; hence, the particle size is uniform (homogeneous). Meanwhile, a PDI value of more than 0.7 indicates an extensive particle size distribution, making the particle size not uniform.27

Table 1. Physicochemical Properties of FB-NPs

Parameters	FB	FB-NPs
Size ± SD (d.nm)	571.9 ± 65.09	222.6 ± 0.87
PDI ± SD	0.56 ± 0.10	0.24 ± 0.02
Note:		
PDI	: Polydispersity index	
SD	: Standard Deviation	
FB	: C. longa-A. paniculate formula	
FB-NPs	: Nano C. longa-A. paniculata	

The characterization of FTIR spectra of FB-NPs showed an absorption at 3358 cm⁻¹, corresponding to the OH stretching group of the FB formula, which overlapped with the -NH stretching of the amine group in carboxymethyl chitosan. The absorption at 2931 cm⁻¹ indicates the presence of the C-H sp³ group from FB. The characteristics of the carboxymethyl chitosan groups were observed at wave numbers 1602 cm⁻¹ (N-H deformation of the primary amine), 1511 cm⁻¹ (asymmetric stretching vibration of COO⁻), and 1030 cm⁻¹ (C-O stretching in -CH₂-OH) (Figure 2). The surface morphology with SEM analysis for FB-NPs. The result of the SEM analysis with magnification 50 X shows that the particle size of FB-NPs is around 344.5 nm and 453.1 nm (Figure 3). The morphology of FB-NPs consists of granules with regular small nanocapsules.

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Figure 2: FTIR Spectrum of Carboxymethyl Chitosan, FB-NPs, and FB

Loading and release of nano C. longa-A. paniculate

The loading of FB-NP bioactive components is determined based on the loading efficiency (%LE) and loading amount (%LA). The loading efficiency of FB-NPs is $35.66 \pm 6.08\%$, which shows the percentage of

FB formula components trapped in the nanocapsule micelles. The loading amount of FB-NPs is $17.83 \pm 3.04\%$, indicating the number of FB formula components trapped in the total nanocapsules. The release percentage of FB-NP bioactive components was tested in the human digestive system at pH 2, 7, and 8.5 for 8 hours (Figure 4). The release is expressed in % release curcumin equivalent. The release of FB-NPs at pH two is observed at $2.89 \pm 0.92\%$, pH seven at $5.78 \pm 1.21\%$, and pH 8.50 at $7.29 \pm 0.95\%$. Poor release at pH two might be due to the aggregation of carboxymethyl chitosan polymers, resulting in bioactive components of FB-NPs being unable to be released. Meanwhile, the best release has been observed in pH 7 and 8.5, indicating that release at pH > 7 is the most optimum pH for release. This condition causes the solubility of carboxymethyl chitosan to increase, which triggers the release of the bioactive components in FB-NPs.



Figure 3: SEM analysis of FB-NPs magnification 50 x

pH Stability of nano C. longa-A. paniculata

Changes in pH affect the stability of the bioactive components in the FB formula and FB-NPs. Stability is observed based on changes in UV-Vis spectra and the turbidity level of the FB and FB-NPs formulas at pH 2-10. Generally, \u03c8max FB at pH 6 (basic pH) is around 345 nm; a hypochromic shift occurs at pH conditions 7-10. Meanwhile, there was a hyperchromic and batochromic shift at pH 5 to 424 nm. In acidic conditions at pH 2-4, FB undergoes hypochromic and batochromic shifts (Figure 5a-b). It indicates that the bioactive component of FB is unstable due to changes in pH. The level of FB turbidity also shows that acidic pH tends to increase, with a decrease in turbidity at pH 2-3 due to the formation of coagulants at the bottom of the bottle. It is suspected that the bioactive components of FB undergo degradation or decomposition into species with lower solubility at acidic pH (Figure 5c). The stability of bioactive components FB-NPs tends to be stable at pH 2-5. While at pH 6-10, there is a hypochromic shift at λ_{max} . The turbidity level of FB-NPs affected by changes in pH shows that pH 2 has the highest turbidity level and decreases with a decrease in pH. The reduction in turbidity level is influenced by the solubility of carboxymethyl chitosan, where changes in pH do not significantly affect the level of tolerance of FB-NPs (Figure 5c). Below pH 5, free -COO- ions will form protonated -COOH. As a result, electrostatic repulsion between chains decreases, allowing the formation of hydrogen bonds, which further increases viscosity. However, deceleration of depolymerization occurs in an alkaline environment, decreasing viscosity.30 This shows that FB-NPs are more able to maintain the stability of bioactive components than FB at pH changes.

Temperature Stability of nano C. longa-A. paniculata

The stability of FB and FB-NPs against temperature changes is observed at 30-100°C. In general, temperature changes do not significantly affect the stability of FB bioactive components. In FB-NPs

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heating at 40-70°C, UV-Vis spectra tend to experience hyperchromic shifts, while at 80-100°C, hypochromic shifts occur.



Figure 4: Release of bioactive components from FB-NPs

It is suspected that heating results in the decomposition of carboxymethyl chitosan, thus affecting the UV-Vis absorption band FB-NPs (Figure 6). Observation of the turbidity level on FB shows that the higher the temperature, the turbidity level decreases, but there is a coagulant (crust) at the bottom of the bottle. It suggests that heating results in the degradation of bioactive components in FB. Conversely, FB-NPs with temperature changes tend to be more stable from the constant turbidity. It aligns with several studies where herbal nanocapsule preparations are relatively stable to temperature changes compared to unencapsulated herbal formulas.^{27,28}



Figure 5: Effect of pH on UV-Vis spectral stability of (a) FB and (b) FB-NPs; (c) Effect of pH on turbidity level FB and FB-NPs

Salt Stability of nano C. longa-A. paniculata Adding salt concentration (NaCl) to FB results in a significant hypochromic shift in the UV-Vis absorption band. In FB-NPs, adding salt concentration tends not to affect bad shifts absorption of UV-Vis (Figure 7). Adding 0.1 M NaCl increases the turbidity level of FB while adding NaCl causes the turbidity level to decrease (Figure 7). However, coagulants (crusts) were formed at the bottom of the bottle. It is



Figure 6: Effect of Temperature on UV-Vis spectral stability of (a) FB and (b) FB-NPs; (c) Effect of Temperature on turbidity level FB and FB-NPs



Figure 7: Effect of Salt concentration on UV-Vis spectral stability of (a) FB and (b) FB-NPs; (c) Effect of Salt concentration on turbidity level FB and FB-NPs

suspected to be caused by the salting-out effect, which results in ionic solubility that decreases the solubility in FB components.³¹ Conversely, in FB-NPs, the addition of salt concentration is directly proportional to the increase in turbidity, although not significant. This suggests that FB

nanoencapsulation with carboxymethyl chitosan can retain the bioactive components of FB.

Toxicity of nano C. longa-A. paniculate

The anticancer activity of the herbal nano formula *C. longa-A. paniculata* (FB-NPs) was carried out using the prestoblue method against breast cancer cells (MCF7 cell line) (Figure 8). The test results showed that FB-NPs had a lower EC₅₀ value compared to FB (free), respectively 13.71 \pm 2.66 µg/mL and 18.43 \pm 6.44 µg/mL. This lower dosage in nanoformulation indicates the efficiency of dosage induced by nano encapsulation. Nanoencapsulation improves drug stability, solubility, bioefficacy, bioavailability, and biodistribution. It can potentially deliver active molecules effectively targeted to specific organs.^{32,33}. Lower dosage also gives benefits of decreasing the accumulation of the drugs during the treatment regimen while still achieving the desired effect.³³ This observation suggests encapsulation of the nanoherbal of *C. longa-A. paniculata* has better dosage efficacy than unencapsulated formula and has potential as an anticancer agent.



Figure 8: Anticancer activity of FB and FB-NPs on MCF-7 cell line

Conclusion

The combination of *Curcuma longa* and *Andrographis paniculata* was successfully nano-encapsulated using carboxymethyl chitosan as the shell material. The encapsulation efficiency and release profile were thoroughly evaluated, revealing that the nano-encapsulated formulation exhibited an optimal pH of 7 for the effective release of its bioactive compounds. Additionally, the encapsulated formulation demonstrated enhanced stability under conditions of high pH, elevated salt concentrations, and increased temperatures. Notably, the nano-encapsulated formulation demonstrated formulation showed significantly improved anticancer activity, achieving an EC₅₀ value of $13.71 \pm 2.66 \,\mu$ g/mL, in comparison to the non-encapsulated formulation. These findings underscore the potential of developing nano-based herbal therapies for cancer treatment in the future.

Conflict of interest

The author reports no conflicts of interest in this work.

Authors' Declaration

The authors hereby declare that the work presented in this article is original and that any liability for claims relating to the content of this article will be borne by them.

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