Tropical Journal of Natural Product Research

Available online at https://www.tjnpr.org

Original Research Article



In-silico Evaluation of Hexagamavunon Analogs for Antibacterial Activity Against Helicobacter pylori

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ARTICLE INFO	ABSTRACT
Article history:	Helicobacter pylori (H. pylori) infection has been associated with gastric cancer. Antibiotic
Received 01 February 2023	resistance has reached dangerous levels. Therefore, finding new anti-effective drugs against H.
Revised 08 August 2023	pylori is crucial. This study evaluated the potential of hexagamavunon (HGV) analogs as anti-H.
Accepted 20 August 2023	pylori drugs through molecular docking. The AutoDock Vina program was used in the
Published online 01 October 2023	molecular tethering process. The ligands (HGV analogs), were docked to the shikimate kinase
	enzyme (PDB ID: 3N2E) and urease (PDB ID: 1E9Y) as the targets in inhibiting <i>H. pylori</i> . The parameter observed was the ligands' binding energy (kcal/mol) compared to native ligands. The results of molecular docking of the shikimate kinase enzyme showed that the binding energies of
Copyright: © 2023 Jannah and Wahid This is an	A6 (-10.7), A7 (-9.9), and A11 (-9.9) were lower compared with native ligand binding energy (-
open-access article distributed under the terms of the	9.8). Also, the binding energy of the urease enzyme with A6 (-7.5), A7 (-8.1), and A11 (-7.7)
Creative Commons Attribution License, which	was lower than the binding energy of the urease with native ligand (-3.4). Low binding energy
permits unrestricted use, distribution, and	correlated with the strength of the bonds between ligands and receptors. HGV analogs, A6, A7,
reproduction in any medium, provided the original	and A11, have higher anti-H. pylori potential than other analogs because they have the lowest
author and source are credited.	binding energies. Further in vitro research is needed to evaluate the potential of HGV analogs as

anti-H. Pylori agents.

Keywords: AutoDock Vina; Binding energy; Molecular docking; Shikimate kinase; Urease

Introduction

Helicobacter pylori (H. pylori), a gram-negative bacteria, is one of the Helicobacter gastric species.¹ H. pylori can infect humans or primates, and it causes chronic gastritis, peptic ulcer, and gastric mucosa-associated lymphoid tissue (MALT).^{2,3} WHO classifies H. pylori infection as the leading cause of gastric cancer.^{4,5} Improper and ineffective treatment of H. pylori infection increases cancer risk.⁶ Antibacterial resistance used in H. pylori treatment has increased, causing a decrease in the effectiveness of therapy.^{4,7,8} The first-line treatment of H. pylori infection is the Proton Pump Inhibitors (PPIs), combined with clarithromycin and amoxicillin or metronidazole for 10-14 days.^{9–11} Inhibiting the synthesis of DNA, proteins, and cell walls is the mechanism of antibiotics used as the primary therapy.¹ The shikimate kinase (SK) enzyme and the urease enzyme are the targets of new anti-H. Pylori agents. SK enzymes play an essential role in synthesising amino acids in H. pylori.^{12–17} Urease enzyme functions as a defence against H. pylori bacteria when living in gastric fluid with low acidity (pH).¹⁸

Curcumin, a compound extracted from *Curcuma longa*, has been investigated for its activity. Curcumin and its derivates have antioxidant, anti-inflammatory, anti-cholesterol, antiviral, anticancer, and chemopreventive activities.^{19–21} Curcumin has shown inhibitory activity against various types of bacteria, including anti-*H. pylori*.^{22–27} The modified structure of curcumin has been synthesized and studied as a mono-carbonyl analog. Modification of the curcumin structure aims to improve its stability and solubility. Introducing different substituents in the phenyl ring of the mono-carbonyl analogs aims to reduce the hydrolytic effect on the molecule.^{28,29}

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Citation: Jannah N and Wahid RAH. *In-silico* Evaluation of Hexagamavunon Analogs for Antibacterial Activity Against *Helicobacter pylori*. Trop J Nat Prod Res. 2023; 7(9):3902-3907 http://www.doi.org/10.26538/tjnpr/v7i9.8

Official Journal of Natural Product Research Group, Faculty of Pharmacy, University of Benin, Benin City, Nigeria

Hexagamavunon (HGV) is a mono-carbonyl analog of curcumin. HGV and its analog have been reported to have activities against gram-positive and gram-negative bacteria, aerobic and anaerobic bacteria, and even have activity against *Mycobacterium tuberculosis*. HGV antibacterial activity studies have been carried out both *in-silico* and *in-vitro*.^{30–34}

In silico studies, known as molecular docking, are carried out by docking a candidate drug molecule with a target receptor to determine drug-protein interactions. Drug-protein interactions can be used as a basis for predicting the candidate molecular activity and molecular affinity.^{35,36} AutoDock Vina is one of the programs used in molecular docking. The advantage of this program is its high speed and accuracy.^{37,38} This research explores molecular docking in determining the activity of HGV analogs as anti-*H. pylori* bacteria by blocking shikimate kinase and urease enzymes.

Materials and Methods

Instruments

The molecular docking used a computer with Intel Celeron N3150 1,60 GHz, ram 2GB, and a Windows 8 operating system. The molecular docking procedure was performed using AutoDock Vina software. AutoDock Tools 1.5.6., Open Babel, Marvin Sketch, and Discovery Studio (DS) Visualizer 2016 were used as supporting software.

Materials

Curcumin analog (Table 1):

A0: 2,6-bis(4-hydroxybenzylidene)cyclohexan-1-one

- A1: 2,6-bis(4-hydroxy-3-methoxybenzylidene)cyclohexan-1-one
- A2: 2,6-di-benzylidenecyclohexan-1-one
- A4: 2,6-bis(4-methoxybenzylidene)cyclohexan-1-one
- A6: 2,6-bis(4-(tert-butyl)benzylidene)cyclohexan-1-one

A7: 2,6-bis(4-(trifluoromethyl)benzylidene)cyclohexan-1-one

A8: 2,6-bis(4-(dimethylamino)benzylidene)cyclohexan-1-one

- A9: 2,6-bis(3,4-dichlorobenzylidene)cyclohexan-1-one
- A10: 2,6-bis(3-chlorobenzylidene)cyclohexan-1-one

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A11: 2,6-bis(4-hydroxy-3,5-dimethylbenzylidene)cyclohexan-1-one A15: 2,6-bis(4-hydroxy-3,5-dimethoxybenzylidene)cyclohexan-1-one A16: 2,6-bis(3,5-dichloro-4-hydroxybenzylidene)cyclohexan-1-one Shikimate kinase enzym (PDB ID: 3N2E) (Figure 1a), urease enzym (PDB ID: 1E9Y) (Figure 1b).

Protein and Ligand Preparation

Shikimate kinase (PDB ID: 3N2E) and urease (PDB ID: 1E9Y) proteins were obtained from the Protein Data Bank website (http://www.rcsb.org). The proteins were prepared (separated from the native ligand and other residues) using DS Visualizer software. The protein was saved in .pdb format. The 3N2E and 1E9Y protein contains three (A, B, C) chains and two (A, B) chains, respectively. The B chain from each protein was used for protein preparation. Native ligand (NL) was separated from the protein and saved in .pdb form. Ligand (A0 – A16) structures were drawn with Marvin Sketch and saved in .pdb form.

Validation of Molecular Docking

The validation procedure was performed to ensure the accuracy of the molecular docking procedure with AutoDock Vina. NL was re-docked to the binding site pocket of its protein. The parameter of the validation procedure was the root mean square deviation (RMSD) in amstrong (Å).

Molecular Docking and Analysis

Ligands (A0 - A16) were docked into the protein's binding pocket. The parameter of the docking procedure was binding energy (kcal/mol) between ligand and protein. The ligand's binding energy was compared with NL's binding energy.

Visualization of Protein-ligand Interaction

The interaction of ligand and protein was observed with DS Visualizer software. This interaction was studied to understand the binding of compounds with amino acid residues of the protein. The output of the docking process was in .pdb format for easy visualization.

Results and Discussion

Protein and Ligand Preparation

Protein preparation was processed by separating the protein from the native ligand and other residues. The B chain from each protein was used as a target in the docking process (Figure 2). The NL of 3N2E is 7-amino-4-hydroxy-3-[(E)-(5-hydroxy-7-sulfonaphthalen-2-

yl)diazenyl]naphthalene-2-sulfonic acid (Figure 3a) while the NL of 1E9Y is acetohydroxamic acid (Figure 3b). The structure of HGV and its analog were acquired from previous studies.²²

Validation of Molecular Docking

NL was re-docked to the binding site pocket of its protein in the validation procedure with RMSD as the parameter. The RMSD of SK and urease was 1.787 Å & 1.689 Å, respectively. SK and urease can be used in molecular docking procedures because of the value of their RMSD < 2 Å.³⁹ In the molecular docking process, ligand and control were docked to the binding site of SK and urease.



Figure 1: Crystal structure of *Helicobacter pylori* (a) shikimate kinase in complex with NSC162535 (PDB ID: 3N2E) (b) urease in complex with acetohydroxamic acid (PDB ID: 1E9Y)

Table 1: Structure of HGV analog ²²					
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Compounds	R ₁	R ₂	R ₃	R4	R ₅
A0	Н	Н	OH	Н	Н
A1	Н	OCH ₃	OH	Н	Н
A2	Н	Н	Н	Н	Н
4	Н	Н	OCH ₃	Н	Н
A6	Н	Н	$t-C_4H_9$	Н	Н
A7	Н	Н	CF ₃	Н	Н
A8	Н	Н	N(CH ₃) ₂	Н	Н
A9	Н	Cl	Cl	Н	Н
A10	Н	Cl	Н	Н	Н
A11	Н	CH ₃	OH	CH_3	Н
A15	Н	OCH ₃	OH	OCH ₃	Н
A16	Н	Cl	OH	Cl	Н

Molecular docking

ISSN 2616-0684 (Print) ISSN 2616-0692 (Electronic)

Determining whether a compound has potential as a drug molecule starts with knowing it meets Lipinski's rule of five. These rules evaluate the physical, chemical, and similarity of compounds with drugs, which can be used to predict their pharmacological and biological activities. Lipinski's rule of five states that an orally active molecule should not violate more than one of the following criteria (1) has a molecular weight of less than 500 Da, (2) logP value of less than 5, (3) the number of hydrogen bond donors is less than 5, and (4) the number of hydrogen bond acceptors less than 10.^{40,41} Previously, all compounds have been selected based on Lipinski's rule of five (RO5). The ligand used in this study meets at least three rules of RO5, while NL of SK only meets two (Table 2). Previous research has confirmed that if a compound does not have more than two RO5, it will affect the solubility and permeability of the molecule.⁴²

The first-line treatment of *H. pylori* infection is the Proton Pump Inhibitor (PPI) group, combined with antibiotics for 10-14 days.^{9-f1} Broad-spectrum first and second-line antibiotics in *H. pylori* therapy have a mechanism of action inhibiting DNA synthesis and cell replication, inhibiting protein, and cell wall synthesis. Future perspective of anti-*H. pylori* drugs are to inhibit virulence factors, metabolism routes, and pH control pathways.^{1,10,12}

The shikimate pathway plays a role in the biosynthetic of aromatic amino acids present in bacteria, fungi, and plants but not in mammals. This pathway consists of seven steps, where a different enzyme catalyzes each step. The enzymes that play a role in this pathway are 3-deoxy-D-arabino-heptulosonate-7-phosphate synthase, 3dehydrogenate synthase, 3-dehydroquinate dehydratase/shikimate dehydrogenase, shikimate kinase, 5- enolpyruvylshikimate-3phosphate synthase, and chorismate synthase.^{17,43} Shikimate kinase is an enzyme that catalyzes the phosphorylation of shikimic acid, the fifth step of the shikimate pathway. This enzyme has become a new target of broad-spectrum anti-H pylori drugs.¹ The result of molecular docking show that the binding energy of SK with ligands A6, A7, and A10 is lower than the binding energy of NL (Table 3). The binding energy of ligands A6, A7, and A10 are -10.7 kcal/mol; -9.9 kcal/mol; and -9.9 kcal/mol, respectively, while the binding energy of NL is -9.8 kcal/mol.

Urease is an enzyme that is responsible for the ability of *H. pylori* to survive in conditions of gastric fluid with low acidity (pH). *H. pylori* produce large amounts of urease in the cytoplasm and on the cell surface. Urease plays a vital role in the hydrolysis reaction of urease to ammonia. The presence of ammonia will neutralize the acidity of the gastric fluid. This enzyme becomes the new target of new broad-spectrum anti-*H. pylori*.^{1,14,18,44} The urease binding energy with all ligands is lower than all the NL (Table 3). The ligands with the lowest binding energy to urease are A6, A7, and A10, which are -7.5 kcal/mol, -8.1 kcal/mol, and -7.7 kcal/mol.

In molecular docking, the parameters observed are the binding energy between the ligand and the target protein. This bond energy shows the stability of the complex between the ligand and protein. Low binding energy values indicate that the ligand-protein complex formed is more stable. The interaction of ligands and proteins, in the form of hydrogen bonds and Van der Waals, has a role in determining the value of binding energy.^{45–50} In this study, A6, A7, and A11 ligands possess more potential to block SK enzymes than other ligands, while all ligands show the potential to inhibit the urease enzyme.

Table 2: The RO5 of ligands

Ligand	Molecular Formula	Molecular Weight (<500 Da)	LogP (<5)	H-Bond Donor (<5)	H-Bond Acceptor (<10)	Violation	Meet RO5
A0	$C_{20}H_{18}O_3$	306.3551	4.99	2	3	0	Yes
A1	$C_{22}H_{22}O_5$	366.4071	4.67	2	5	0	Yes
A2	$C_{20}H_{18}O$	274.3563	5.6	0	1	1	Yes
A4	$C_{22}H_{22}O_3$	334.4083	5.28	0	3	1	Yes
A6	C ₂₈ H ₃₄ O	386.569	8.69	0	1	1	Yes
A7	$C_{22}H_{16}F_6O$	401.3523	7.35	0	7	1	Yes
A8	$C_{24}H_{28}N_2O$	360.4919	5.81	0	3	1	Yes
A9	$C_{20}H_{14}Cl_4O$	412.137	8.01	0	5	1	Yes
A10	$C_{20}H_{16}Cl_2O$	343.246	6.81	0	3	1	Yes
A11	$C_{24}H_{26}O_3$	362.4614	7.04	2	3	1	Yes
A15	$C_{24}H_{26}O_7$	426.459	4.36	2	7	0	Yes
A16	$C_{20}H_{14}Cl_4O_3$	444.135	7.41	2	7	1	Yes

	Table 3: Molecular	docking resul	t of shikimate kinase	and urease
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	Shikimate Kinas	e (SK)		Urease		
Ligand	RMSD (Å)	Binding (kcal/mol)	Energy	RMSD (Å)	Binding (kcal/mol)	Energy
NL	1.787	-9.8		1.689	-3.4	
A0	1.390	-7.4		0.675	-6.5	
A1	1.203	-7.3		1.333	-7.4	
A2	1.535	-7.4		1.752	-6.5	
A4	0.566	-8.1		0.903	-6.7	
A6	0.128	-10.7		1.338	-7.5	
A7	0.910	-9.9		0.254	-8.1	
A8	1.129	-7.9		0.334	-7.0	
A9	1.594	-8.9		1.395	-7.4	

ISSN 2616-0684 (Print) ISSN 2616-0692 (Electronic)

A10	1.160	-8.5	1.136	-6.9
A11	0.875	-9.9	1.381	-7.7
A15	0.263	-8.3	1.497	-7.0
A16	0.267	-9.0	1.866	-7.1



Figure 2: Structure of B chain of (a) shikimate kinase (b) urease



Figure 3: Structure of (a) 7-amino-4-hydroxy-3-[(E)-(5-hydroxy-7-sulfonaphthalen-2-yl)diazenyl]naphthalene-2-sulfonic acid (b) acetohydroxamic acid

Interaction of Protein-ligand

The ligand-protein interactions show that A6, A7, and A11 form bonds with the amino acid residue of SK, which is the binding site of the native ligand (Table 4). Ligands A6 and A11 interact with 5 of the same amino acids, ARG116, ARG132, MET10, PHE48, and PHE9, while A7 interacts with 3 of the same amino acids, ARG116, PHE48, and PHE9. Ligands A7 and A11 form 2 hydrogen bonds with SK proteins with a distance of 3.75 & 3.31 Å (A7) and 2.86 & 3.25 Å (A11). A6 ligands form 3 hydrogen bonds at 2.85, 4.12, and 3.28 Å. The results of ligand-protein interactions on urease showed no similarity between amino acids bound to the original ligand and the test ligand (Table 4). Native ligands form bonds with two amino acids (ALA365; ASP362) and three nickel metals. A6 ligand interacts with four amino acid residues and forms one hydrogen bond with a distance of 3.14 Å. A7 ligands interact with nine amino acid residues of urease, four forming hydrogen bonds with a range of 3.15, 3.92, 3.17 and 3.20 Å. A11 ligands interact with seven amino acid residues, and three form hydrogen bonds with a distance of 4.15, 3.23, and 3.29 Å.

Interactions between proteins and ligands occur in various types of bonds. Bonds that have a significant role in determining the binding energy of ligand-protein are the hydrogen and Van der Waals bonds. The number and distance of hydrogen and Van der Waals bonds determine the low strength of the ligand-protein binding energy.^{45–51} In SK interactions with ligands, A6 shows the lowest binding energy because it has more hydrogen bonds and shorter distances than other ligands.

Conclusion

This paper reported the HGV analog employed in the *in silico* investigation of SK and urease enzymes in *H. pylori* bacteria. Analogs A6, A7, and A11 showed the highest potential to inhibit *H. pylori* bacteria by inhibiting the SK and urease enzymes from the molecular docking results. However, further *in vitro* studies are necessary to validate the potential of HGV analogs as anti-*H. Pylori* drugs.

Conflict of Interest

The authors declare no conflict of interest.

Authors' Declaration

The authors hereby declare that the work presented in this article is original and that any liability for claims relating to the content of this article will be borne by them.

Acknowledgements

We thank the Pharmacy Department, University PGRI of Yogyakarta and all parties for their research assistance in this paper.

Licond	Amino acid involved in the interaction						
Ligand	Shikimate Kinase (SK)	Urese					
NI	ARG116; ARG132; ARG57; GLY80; MET10; PHE48;	ALA365; ASP362; HIS248; NI3001; NI3001;					
INL.	PHE9	NI3002					
4.0	ARG45; ASP33; LYS111; LYS14; MET10; PHE48;	ALA185; MET182; PRO142; TYR189					
AU	SER15; VAL44						
	ARG116; LYS115; PHE48; PRO117	ALA563 ; ASN309; ASN558; GLN378; GLN564;					
A1		GLU371; SER151; SER567; VAL444; VAL560					
Δ2	ARG116; MET10; PHE48; VAL44	ALA563; GLN378; GLU371; GLN564; THR307;					
112		VAL560					

Table 4: Interaction of ligand-protein of shikimate kinase and urease

Ligand	Amino acid involved in the interaction			
Liganu	Shikimate Kinase (SK)	Urese		
	ASP33; GLY80; LYS115; MET10; PHE48	ALA563; ARG375; GLU371; SER151; THR307;		
A4		VAL560		
٨	ARG116; ARG132; LEU119; LYS115; MET10; PHE48;	ALA563; GLU371; PHE569; VAL369		
Au	PHE9			
Δ.7	ALA125; ARG116; ARG45; ILE105; LEU119; PHE101;	ALA557; ALA563; ASN309; ASN558; GLN378;		
A/	PHE129; PHE48; PHE9	GLU371; LYS559; SER151; VAL560		
A8	ARG45; LYS115; MET10; PHE48; VAL44	ALA563; ARG368; GLN378; GLU371; PHE569		
40	ARG116; ILE105; LEU104; LEU108; LEU119; MET10;	ALA563; GLN378; GLU371; VAL560		
A9	PHE48; PHE9; VAL44			
A10	ARG116; ARG132; GLY80; LYS115; MET10; MET84;	ALA563; GLN378; GLU371; LYS445; LYS559;		
	PHE48; PHE9; TYR136; VAL83	SER567; VAL560		
A11	ARG116; ARG132; LEU104; LEU108; LEU119;	ALA563; GLN378; GLU371; GLY370; SER151;		
	LEU128; MET10; PHE48; PHE9	VAL444; VAL560		
A15	ARG116; ARG45; ARG57; GLU53; GLY80; GLY81;	ALA563; ASN309; GLN564; GLU371 ; PHE569;		
	LYS115; LYS115; MET10; PHE48	SER151		
A16	ARG116; ARG57; ASP33; GLY80; LYS115; MET10;	ALA563; GLU371; LEU562; LYS445; PRO305;		
	PHE48; VAL44	SER151; VAL560		

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